



THE BENEFITS OF DIRECT MERCURY ANALYSIS: A TECHNOLOGY REVIEW

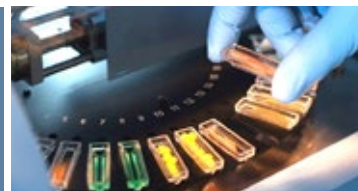


The analytical challenges in mercury determination related to tedious sample preparation process and memory effect using latest generation of ICP-MS or CVAA/AF systems can be overcome by Direct Mercury Determination (DMA-80) system.

INTRODUCTION

The inherent problem of old vapor atomic absorption/atomic fluorescence (CVAA/AF), or Inductively Coupled Plasma Mass Spectrometry (ICP-MS) lies in the fact that all these techniques are solution based, which means that if the sample is not a liquid it has to be digested before it is introduced to the instrument. Because of the high volatility of mercury, this can lead to loss of analyte and poor recoveries if the sample is not digested correctly. In addition, even if the sample is a liquid, mercury is notorious for its memory effects, particularly using techniques such as ICP-OES or ICP-MS because it can stick to the walls of the sample container as well as surfaces of the sample introduction system. To compensate for this, long washout times

are typically required to ensure the previous sample is not still resident in the sample line. In some cases, it requires the addition of a few ppm of gold, which acts as a strong oxidizing agent to convert the mercury to the mercuric ion to stop it from being reduced to elemental mercury and lost to the atmosphere or absorbed into the sample line. This is further compounded if the solutions are left for extended periods of time before they are analyzed. The bottom line is when carrying out mercury measurements at low levels, great care must be taken to ensure the data generated is representative of the mercury in the original sample.



| DIRECT MERCURY ANALYSIS

One of the most widely accepted ways around these limitations is to carry out the measurement by direct mercury analysis, which is a technique used for the determination of total mercury directly in solid, liquid and gas samples, using the principle of thermal decomposition, amalgamation and atomic absorption. In this approach, a decomposition furnace is used to release mercury vapor instead of the chemical reduction step used in traditional liquid-based analyzers. Both solid and liquid matrices can be loaded onto the instrument's autosampler and analyzed without acid digestion or sample preparation prior to analysis. Because this approach does not require the conversion of mercury to mercuric ions, lengthy sample pretreatment steps are unnecessary. As a result, there is no need for reagents such as highly corrosive acids, strong oxidizing agents or reducing chemicals, which means no hazardous waste to be disposed of. Direct mercury analysis is a well-established analytical technique used by

the environmental, biological, clinical, food, industrial and academic communities that has been approved by testing/standards organizations such as the EPA in method 7473 (1) and the ASTM using Method D6722 (2).

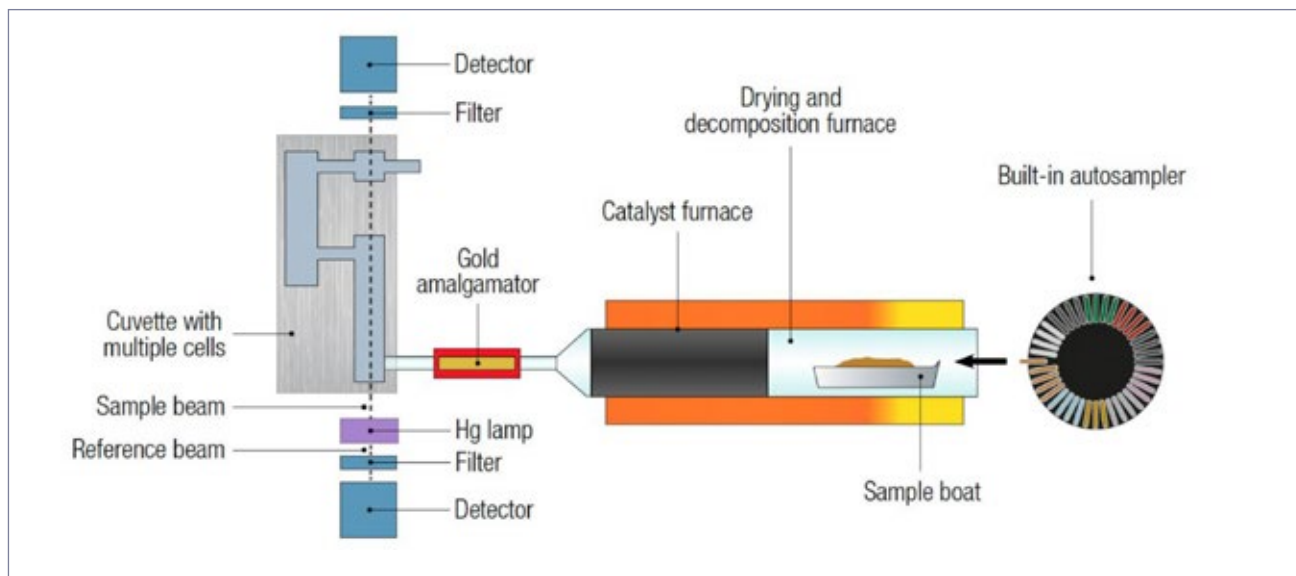
| PRINCIPLE OF OPERATIONS

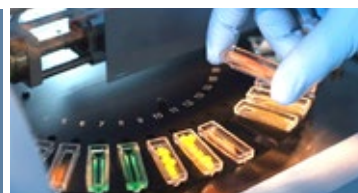
Direct mercury analysis can be broken down into four distinct steps:

- Thermal decomposition
- Catalytic conversion
- Amalgamation
- Atomic absorption detection

Controlled heating stages are implemented to first dry and then thermally decompose a sample introduced into a quartz tube. A continuous flow of air or oxygen carries the decomposition products through a hot catalyst bed where halogens, nitrogen, and sulfur oxides are trapped, allowing for the analysis of even reactive or flammable samples. All mercury species are reduced to elemental mercury (Hg⁰) and are then carried along

Figure 1: A schematic of the DMA-80 *evo* direct mercury analyzer.





with reaction gases to a gold amalgamator where the mercury is selectively trapped. All non-mercury vapors and decomposition products are flushed from the system by the continuous flow of gas. The amalgamator is subsequently heated and releases all trapped mercury to the single- or double-beam (depending on model), fixed wavelength atomic absorption spectrophotometer. Absorbance is measured at 253.7 nm as a function of mercury content. Since mercury is thermally released from the sample and pre-accumulated, measurements are matrix-independent and instrument calibrations are long-lasting. A schematic of a direct mercury analyzer (DMA-80 *evo*- Milestone) is shown in Figure 1.

| DOUBLE BEAM TECHNOLOGY: THE NEW DMA-80 *evo*

The DMA-80 was initially developed using single-beam AA technology. In this technique, a beam of light (photons) from the hollow cathode light source is directed through the sample compartment, where the analyte atoms generated by the excitation source absorb a characteristic wavelength of light, which is then passed into the optical system, detected and converted into an electrical signal.

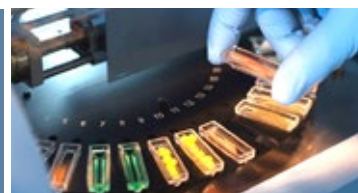
Since variations in the lamp energy can potentially compromise precision and detection capability, for the very first time, the benefits of double-beam technology have now been incorporated into the design of a direct mercury analyzer with the Milestone DMA-80 *evo*. With this technology, a reference beam monitors the lamp energy whereas the sample beam reflects absorption of the analyte photons. The observed absorbance measurement is the ratio of the sample and reference beams. Double-beam technology

compensates for the effects due to drift in lamp intensity, electronic and mechanical fluctuations, and thermal instability that affect both the sample and reference beams equally. This significantly increases the signal-to-noise ratio, which greatly improves the stability of the signal, resulting in a lower limit of quantification and enabling more accurate and precise mercury determinations even at the low-ppt level.

| AUTO-BLANK FEATURE

In addition to the unique double-beam technology, the DMA-80 *evo* also incorporates software which recognizes the mercury level of a sample and, in the case of high concentrations, introduces blank cycles in the sample run to clean the system prior to running the next one. After this cycle, the instrument evaluates the blank level and proceeds automatically to the next sample. For example, if the Hg level is above 10 ng as previously determined by the user, the software will start blank cycles to clean the system prior to processing the next sample. Then, the blank cycles are carried out until the Hg level is below 0.5 ng (using a maximum of 2 blank cycles). At this point, the analysis will continue. If the blank cycle value is still over 0.5 ng, the operator can abort at any time and the Auto Blank feature will stop. Both the mercury limits and the amount of blank cycles can be modified.

This process is demonstrated in Figure 2, which shows the measurement of two fish samples (swordfish and sea bass) of 8 and 118 ng, respectively, that are automatically followed by two auto blanks until the desired blank value of 0.5 ng is reached. A TORT 3 lobster CRM is analyzed at the end of the run, showing good agreement with the certi-

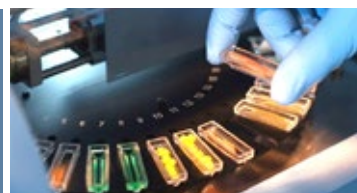


fied value of 288 µg/kg. The atomic absorption peak profiles of samples, blanks and CRM are shown in the far right hand column for confirmation purposes. This Auto Blank

feature, which can be customized based on mercury levels, therefore allows for the highest accuracy particularly at trace levels.

Figure 2: Data showing the DMA-80 evo Auto Blank feature.

Sample ID	Hg (ng)	Result µg/Kg	Certified Value µg/Kg	Graphic
Swordfish	8.03	71.13		
Seabass	117.76	1239.66	-	
Auto-Blank (1)	0.53	-	-	
Auto-Blank (2)	0.25	-	-	
Lobster Hepatopancreas (TORT-3)	10.24	287.65	292 ± 22	



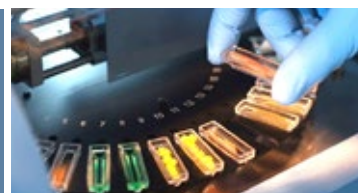
PRODUCTIVITY

The DMA-80 *evo* offers a significant enhancement in productivity over ICP-MS and cold vapor systems. Not only do these techniques require labor-intensive and time-consuming sample preparation steps, which can lead to mercury losses, but also all the hazardous waste chemicals and solutions must be disposed of in a safe manner. In addition, ICP-MS and ICP-OES are very prone to mercury carry over, as mentioned previously. To avoid this problem the user typically has to dilute the sample, degrading the detection capability. In extreme ca-

ses with high levels of contamination, users have to clean the instrument using a gold solution or, in a worst case scenario, replace the sample introduction components. All of these measures have a significant impact on sample throughput and productivity. The enhancement in productivity of the DMA-80 *evo* over CVAA is demonstrated in Table 1, which shows a factor of 4x difference in the cost per sample. This translates into total savings of almost \$500 per day or \$10,000 per month.

Table 1: Productivity comparison of DMA-80 *evo* with CVAA.

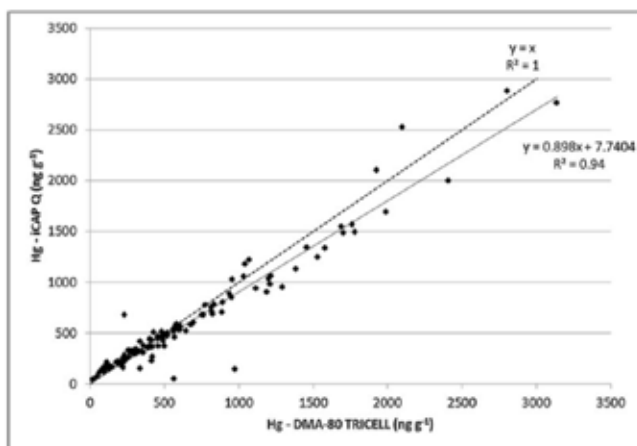
	DMA-80 <i>evo</i>	CV-AA
Calibration during analysis	63\$ / DORM3 <i>Valid for thousands of samples</i>	62.37\$ /EPA Method 7471B <i>Daily calibration is required</i>
Time per sample <i>See table 1</i>	6 minutes	12 minutes
Daily sample analysis	80 samples	40 samples
Labor cost <i>Based on 8 hour working day*</i>	160\$	160\$
Labor cost per sample	2\$	4\$
Reagent cost per sample	0.30\$	4.50\$
Reagent cost per day <i>Based on 80 samples</i>	24\$	180\$
Total cost <i>Labor cost + calibration + reagents</i>	273\$	562.37\$
Total cost per sample	2.30\$	8.50\$



A COMPARISON WITH ICP-MS

A recent study by Domanico and co-workers at the University of Calabria in Italy compared thermal decomposition atomic absorption (DMA-80) with quadrupole-based ICP-MS (iCAP Q) for the quantitation of Hg in human hair (3). After the washing procedure to minimize the external contamination, two aliquots were taken from each hair sample. The first was used for direct analysis of Hg using the DMA-80 and the second was digested for Hg determination by ICP-MS. Ini-

Figure 3: Correlation between quadrupole ICP-MS and the DMA-80 for the determination of Hg in hair (3)

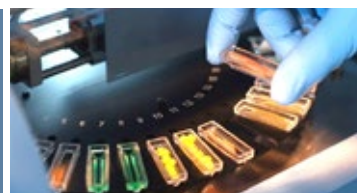


tial results indicated that the two data sets were fully comparable and were not statistically different. The two techniques demonstrated results with a good coefficient of determination ($R^2 = 0.94$) despite differences in dynamic ranges and method detection limits. Both techniques satisfied internal performance requirements and method validation resulting in sensitive, precise and reliable results as demonstrated in Figure 3. However, the authors further concluded that the DMA-80 offered an attractive alternative to ICP-MS for hair analysis because of its re-

duced sample handling, lower risk of contamination due to the absence of the digestion step, less time-consuming operation, and higher sample throughput, which results in a more cost-effective analysis. Let's now take a closer look at some routine applications, which have been traditionally addressed by cold vapor AA/AF or ICP-MS.

ENVIRONMENTAL ANALYSIS

As more emphasis is placed on the monitoring of mercury emissions from power plants, there is an increasing demand to characterize soil, sediments and waste waters for remediation purposes. Several methods including ICP-MS and cold vapor are available for mercury analysis of these types of samples. However, both these techniques require complex sample preparation procedures that are labor intensive, time-consuming and subsequently expensive. As a result, direct mercury analysis, as described in EPA Method 7473, was specifically developed for the rapid determination of mercury according to the Resource Conservation and Recovery Act (RCRA) governing the management of hazardous waste. Therefore, to exemplify its real-world capabilities the DMA-80 *evo* was used to carry out mercury determinations on a suite of soils, sludges and waste waters. In addition, two certified reference materials (NIST 2709 San Joaquin Soil CRM and BCR 277 Estuarine River Sediment CRM) were both analyzed at multiple times during the analytical run for quality assurance/control purposes. A subset of these data is exemplified in Table 2, which shows five separate analyses of the soil and estuarine sediment CRMs, together with the recovery and precision of a waste water sample spiked with 1 $\mu\text{g/L}$ Hg. It can be clearly seen that the re-



sults and statistical data of both CRMs were in very good agreement with the certificate values, which were analyzed by cold vapor isotope dilution (CV-ID) ICP-MS, recognized as being one of the most precise techniques for mercury quantitation. It should also be emphasized that with the wastewater sample, a calibration blank was analyzed before and after the five wastewater samples and 1 $\mu\text{g/L}$ spike, showing the capability of the double-beam system to achieve very low background levels even after a suite of complex, real-world samples.

| PETROCHEMICAL, OIL, CEMENT AND ENERGY INDUSTRY

In the refining of crude oil, mercury is not only released into the environment during the refining process, but also from the combustion of refinery products such as gasoline and diesel fuel. Although mercury levels are typically less than 2 $\mu\text{g/kg}$ in most crude oils, it has the potential to accumulate in equipment and cause operational problems in refining plants. It is therefore critical to monitor mercury to ensure operators are not exposed to the environmental impact of the mercury. As a result, two analytical methods developed for the measurement of mercury in petrochemical samples are UOP 938-10 (Universal Oil Products) for total mercury and mercury species in liquid hydrocarbons (4) and the more recent ASTM D7623-10, a test method for total mercury in crude oil using combustion-gold amalgamation and cold vapor atomic absorption (5). However it is important to emphasize that if the quantitation of mercury species is not required, the preferred approach is the ASTM method.

The cement industry is another segment that requires mercury monitoring. In their manufacturing process, limestone and other mi-

neral components such as alumino-silicates and clays are fused together at high temperatures to form a clinker-type material, which is then ground up and mixed with other additives to make cement. The heat required by this production process is typically achieved by burning coal or coke (a type of charcoal). However, due to high levels of mercury both in geological raw materials and fossil fuels, the production of cement clinker can potentially result in significant levels of mercury emission into the atmosphere. In fact, it is estimated that the total amount of mercury generated by the electricity and cement industries exceeds 60 tons per year combined (6).

For this reason, there is a growing demand for rapid and easy-to-use mercury analysis procedures. Direct mercury analysis is therefore a cost-effective alternative to traditional methods and has been used successfully to determine total mercury in a large variety of samples related to the generation of electricity and the production of cement. A typical data set of these types of samples is shown in Table 3, which includes three different coke, coal, and limestone samples analyzed consecutively with blank levels in between. It can be seen quite clearly that the samples cover a wide range of mercury levels, from $<1 \mu\text{g/kg}$ in the coke sample, to approximately 10 $\mu\text{g/kg}$ in the limestone, and up to about 30 $\mu\text{g/kg}$ in the coal sample. Two points should be emphasized. First, blank samples were analyzed at the beginning of the analysis, between each sample matrix and at the end of the analytical run. The second point is that NIST 2691 coal fly ash CRM was analyzed at the very end and achieved good accuracy compared to the certificate value, where the mercury was quantitated using CV-ID ICP-MS.

TECHNOLOGY REPORT

DMA-80 *evo*

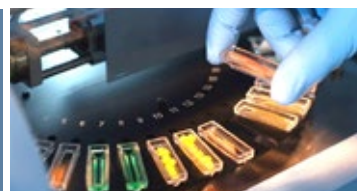


Table 2: Mercury results for NIST 2709 San Joaquin soil and BCR 277R Estuarine Sediment CRMs, together with the analysis of a wastewater sample spiked with 1 µg/L using the DMA-80 *evo*.

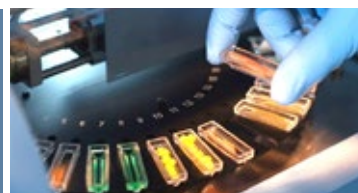
NIST 2709 San Joaquin Soil CRM	DMA Concentration (mg/kg)
NIST 2709 Sample 1	1.378
NIST 2709 Sample 2	1.404
NIST 2709 Sample 3	1.406
NIST 2709 Sample 4	1.404
NIST 2709 Sample 5	1.424
Mean	1.400
SD	0.016
RSD (%)	1.17
NIST 2709 Certificate Value	1.40 ± 0.08
BCR 277R Estuarine Sediment CRM	DMA Concentration (mg/kg)
BCR 277R Sample 1	0.125
BCR 277R Sample 2	0.122
BCR 277R Sample 3	0.123
BCR 277R Sample 4	0.127
BCR 277R Sample 5	0.127
Mean	0.124
SD	0.002
RSD (%)	1.83
BCR 277R Certificate Value	0.128 ± 0.017
Wastewater Sample	DMA Concentration (µg/L)
Blank 1	0.0001
Wastewater Sample 1	0.105
Wastewater Sample 2	0.108
Wastewater Sample 3	0.109
Wastewater Sample 4	0.106
Wastewater Sample 5	0.105
Wastewater Sample Mean/SD	0.107 ± 0.002
Wastewater Sample 5 + 1 µg/L spike	1.12
% Spike Recovery (1 µg/L)	101%
Blank 2	0.0001

Table 3: Three different sample matrices (coke, coal and limestone) analyzed consecutively with the DMA-80 *evo*, with blank levels in between and a NIST 2691 coal fly ash CRM analyzed at the end of the run.

Sample	Weight (g)	DMA Concentration (µg/kg)
Blank	-	0.0011
Coke	0.059	0.83
Coke	0.052	0.90
Coke	0.046	0.79
Blank	-	0.0051
Coal	0.060	32.9
Coal	0.059	29.1
Coal	0.062	30.5
Blank	-	0.0037
Limestone	0.109	10.2
Limestone	0.109	10.2
Limestone	0.109	9.96
Blank	-	0.0023
NIST 2691 CRM	0.105	59.1
NIST 2691 Certified Value		57.8 ± 4.3

TECHNOLOGY REPORT

DMA-80 *evo*



| CONCLUSIONS

With over 2500 systems installed worldwide, direct mercury analysis has proved itself to be a viable complementary technique to CVAA and ICP-MS for carrying out mercury determinations. Its ability to analyze solids directly with minimal sample handling is very attractive for remote field-based studies as well as for labs that don't have sophisticated sample preparation and digestion procedures available. In addition, the ability to introduce the sample directly into the instrument means contamination issues are minimized, which makes it ideally suited for non-experienced operators who are not versed in the finer points of analytical chemistry. And the DMA-80 *evo* with its new double-beam light path enhances its capability even further by compensating for any thermal and mechanical instability in the light source and optical components. Moreover, the Auto Blank feature goes a long way to ensuring that any abnormally high concentration samples do not negatively affect low samples that come immediately after it in the sample run. With the recognized industry-leading application and service support of Milestone Inc. behind it, there is no question that the new DMA-80 *evo* is well-placed to take rapid, cost-effective and high productivity mercury determinations to the next level.

| ABOUT MILESTONE

With over 50 patents and more than 20,000 instruments installed in laboratories around the world, Milestone has been widely recognized as the global leader in metals prep technology for the past 30 years. Committed to providing safe, reliable and flexible platforms to enhance your lab's productivity, customers worldwide look to Milestone for their metals digestion, organic extractions, mercury analysis and clean chemistry processing needs.

| REFERENCES

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6. Clean Air Act, 42 U.S.C. §7401 et seq., (1970); <https://www.epa.gov/laws-regulations/summary-clean-air-act>



Robert Thomas

is the principal of Scientific Solutions, a consulting company that serves the application and writing needs of the trace element user community. He has worked in the field of atomic and mass spectroscopy for more than 40 years and has written over 90 technical publications including a 15-part tutorial series on ICPMS. He recently completed his fourth textbook entitled Measuring Elemental Impurities in Pharmaceuticals: A Practical Guide. He has an advanced degree in analytical chemistry from the University of Wales, UK, and is also a Fellow of the Royal Society of Chemistry (FRSC) and a Chartered Chemist (CCChem). He has led the heavy metals, plasma spectrochemistry task force on the ACS Committee on Analytical Reagents for the past 18 years.



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